PennState Research Experience for Undergraduates



# Optimized Growth Conditions of MoS<sub>2</sub> Nanoflowers for HER

**Evolution** 

Ricardo Ortiz Cisneros<sup>1</sup>, Nicholas Simonson<sup>2</sup>, and Dr. Joshua A. Robinson<sup>2</sup>

<sup>1</sup>Department of Chemistry and Biochemistry, California State University, Los Angeles; <sup>2</sup>Department of Material Science and Engineering, Penn State University

## Introduction

Climate change induced by anthropomorphic activity has driven the search for sustainable sources of energy. Many of these technologies are limited by the slow reaction kinetics of the oxygen reduction reaction (ORR) and hydrogen evolution reaction (HER) which demand expensive metal catalysts for practical use. Herein, inexpensive MoS<sub>2</sub> nanoflowers are grown and have the potential to catalyze many of the reactions required for future sustainable energy applications.



Cal

State

**Figure 1**: An example of future sustainable energy: a paper based direct formate fuel cell dependent on the oxidation of potassium formate<sup>1</sup>.

**Figure 2**: Schematic of the furnace used to grow the nanoflowers<sup>2</sup>



#### Methods

The  $MoS_2$  nanoflowers were grown via powder vapor disposition (PVD) from precursor molybdenum oxide and sulfur chips on flexible graphite paper. Various conditions were investigated for optimal growth including argon flow, temperature, time, and pressure.





## Results

**Figure 3**: (left to right) The graphite paper substrate before and after the growth. The purple color indicates the growth of the flowers.



```
Figure 4: FESEM surface image of the nanoflowers
```

**Figure 5**: Raman spectra of the substrate surface after growth. The  $E_{2g}$  and  $A_{1g}$  vibrational modes of  $MoS_2$  at 380 and 406 cm<sup>-1</sup>, respectively show high quality  $MoS_2$  growth.

## **Discussion and Conclusion**

Figures 3-5 suggest the presence of  $MoS_2$  nanoflowers. Optimal growth seems to occur at conditions of 200torr, 200sccm, 800°C, and 30min for pressure, flow rate, temperature and bake length, respectively. Given the system drift inherit in PVD techniques, these conditions vary over time even when using the same

equipment.



Owing to the increased edge sites of the  $MoS_2$  nanoflowers, these substrates are expected to catalyze various reactions. Electrochemical techniques such as cyclic voltammetry and linear sweep voltammetry should be deployed to conduct the catalytic characterizations of these nanostructures.

#### References

 Galvan, V.; Domalaon, K.; Tang, C.; Sotez, S.; Mendez, A.; Jalali-Heravi, M.; Purohit, K.; Pham, L.; Haan, J. Gomez, F. A., An improved alkaline direct formate paper microfluidic fuel cell. *Electrophoresis* 2015, 37 (3), 504-510.
Bhimanapati, G. R.; Hankins, T.; Lei, Y.; Vila, R. A.; Fuller, I.; Terrones, M.; Robinson, J. A., Growth and Tunable Surface Wettability of Vertical MoS2 Layers for Improved Hydrogen Evolution Reaction. *ACS Applied Materials and Interfaces* 2016, 8, 22190-22195.

The Penn State REU program in Interdisciplinary Materials and Materials Physics is supported by the Penn State Department of Physics and the Center for Nanoscale Science (NSF-MRSEC) and the National Science Foundation (DMR1460920 and DMR1523588).